

Concept Review Section Changes Of State Answers

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Concept Review Section: Changes of State 1. Identify the process that causes each change of state described below.

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The change in the internal energy of a system is equal to the difference between the internal energy of the final and initial states. According to the law of conservation of energy, energy can be neither created nor destroyed, but it can be transferred to or from the surroundings.

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Skills Worksheet. Concept Review. Section: Circuits. 1. Identify the types of elements in the schematic diagram shown below. Give the number of each type of element in the diagram. 2. Predict which fuse would give the greatest protection against high current in a circuit—a 20 A fuse or a 30 A fuse.

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Concept Review - Manchester High School
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Concept Review: Energy. 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed. 9. Heat is the energy transferred between objects that are at different temperatures.

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Concept Review with Key Terms - Pearson Education
Concept Review j j. 1. A chemical is any substance that has [de]nit,e, indefinite) composition. Changes in chemicals, or chemical reactions, take place [only in test tubes, all 2. The type and arrangement of [particles, crystals] in a sample of matter determine the properties of the matter.

Concept Review Section Changes Of
Identify two changes of state that require energy to be used, and two changes of state that require the release of energy. Evaporation and sublimation require energy to be used. Condensation and freezing release energy. Apply the law of conservation of mass to explain how mass is conserved when water evaporates.

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